L Numbe	r Hits	Search Text	DB	Time stamp
9	264	560/218.ccls.	USPAT;	2004/02/26 05:48
	,		US-PGPUB;	
			EPO; JPO;	
			DERWENT	
10	388	562/600.ccls.	USPAT;	2004/02/26 05:48
			US-PGPUB;	• • • • • • • • • • • • • • • • • • • •
		3	EPO; JPO;	
			DERWENT	
11	0	5296205.URPN.	USPAT	2004/02/26 05:48
12	Ö		USPAT	2004/02/26 05:48
13	475		USPAT;	2004/02/26 05:48
13	1,3	201/ 57:0015:	US-PGPUB;	2001, 02, 20 03.10
			EPO; JPO;	
		· ·	DERWENT	·
14	627	560/218.ccls. or 562/600.ccls.	USPAT;	2004/02/26 08:47
14	027	500/210.ccis.	US-PGPUB;	2004/02/20 08.47
				***
			EPO; JPO;	
	2.5	man and a district the second second	DERWENT	0004/00/05 30 56
15	319	"gas-injection"	USPAT;	2004/02/26 12:56
			US-PGPUB;	
		,	EPO; JPO;	
			DERWENT	1
17	6821	"beam type"	USPAT;	2004/02/26 05:49
			US-PGPUB;	
	•		EPO; JPO;	
			DERWENT	
18	0	(560/218.ccls. or 562/600.ccls.) and	USPAT;	2004/02/26 05:49
		"gas-injection"	US-PGPUB;	
			EPO; JPO;	
		*	DERWENT	
19	0	(560/218.ccls. or 562/600.ccls.) and "beam	USPAT;	2004/02/26 05:49
		type"	US-PGPUB;	,,
		of po	EPO; JPO;	
			DERWENT	
20	0	261/97.ccls. and "gas-injection"	USPAT;	2004/02/26 05:49
20		201/97.ccis. and gas injection	US-PGPUB;	2004/02/20 03:45
		÷	EPO; JPO;	
			DERWENT	
21	105024	gupport add plate		2004/02/26 05:49
21	105834	support adj plate	USPAT;	2004/02/26 05:49
			US-PGPUB;	• .
			EPO; JPO;	
2.2	242		DERWENT	2004/00/05 05 40
22	242	"beam type" and (support adj plate)	USPAT;	2004/02/26 05:49
			US-PGPUB;	
			EPO; JPO;	
			DERWENT	
23	0	"beam type" near4 (support adj plate)	USPAT;	2004/02/26 05:49
			US-PGPUB;	
			EPO; JPO;	
			DERWENT	
27	1737	RMax	USPAT	2004/02/26 09:13
28	0	("gas-injection" near4 (support adj plate))	USPAT;	2004/02/26 05:49
		and RMax	US-PGPUB;	
			EPO; JPO;	
		, ,	DERWENT	
1	2	US-5296205-\$.DID.	USPAT;	2004/02/26 05:49
			US-PGPUB;	
	3		EPO; JPO;	
			DERWENT	
30	0	261/97.ccls. and RMax	USPAT;	2004/02/26 08:49
50			US-PGPUB;	= 00.00.45
		•	EPO; JPO;	
			DERWENT	
2	_		USPAT	2004/02/26 05:49
2	1 1		1	
3	1		USPAT	2004/02/26 05:49
4	1		USPAT	2004/02/26 05:49
5	1		USPAT	2004/02/26 05:49
6	1		USPAT	2004/02/26 05:49
7	1		USPAT	2004/02/26 05:49 2004/02/26 05:49
8			USPAT	1 000 1 100 100 00 10

16		2	261/97.ccls. and (560/218.ccls. or	USPAT;	2004/02/26 05:49
			562/600.ccls.)	US-PGPUB;	
			302/000.0013.7	· ·	1
	İ			EPO; JPO;	
1				DERWENT	
24		12	"beam type" same (support adj plate)	USPAT;	2004/02/26 06:20
			boam offer same (bappore and process	US-PGPUB;	,,
	j		2	1	
				EPO; JPO;	
				DERWENT	
25		2	"gas-injection" near4 (support adj plate)	USPAT;	2004/02/26 05:49
25	1	Z	gas-injection heart (support adj prace)	1 '	2001/02/20 03:19
				US-PGPUB;	
ļ				EPO; JPO;	
				DERWENT	
0.5		4	4814117.URPN.	USPAT	2004/02/26 05:49
26		4			
29	-	1	(560/218.ccls. or 562/600.ccls.) and RMax	USPAT	2004/02/26 05:49
31		1	•	USPAT	2004/02/26 05:54
32		1		USPAT	2004/02/26 05:54
33		1		USPAT	2004/02/26 05:54
34		1		USPAT	2004/02/26 05:54
35		1		USPAT	2004/02/26 05:55
1					2004/02/26 05:55
36	į	1		USPAT	
37	×	1		USPAT	2004/02/26 05:55
38		1		USPAT	2004/02/26 05:56
39		1		USPAT	2004/02/26 05:56
1		_			
40		. 1		USPAT	2004/02/26 05:57
41		8	"gas-injection" and (support adj plate)	USPAT;	2004/02/26 06:32
1				US-PGPUB;	
1				EPO; JPO;	e e
			ĺ		
1.			•	DERWENT	1
42		5	4576763.URPN.	USPAT	2004/02/26 06:23
43		1		USPAT	2004/02/26 06:25
44		1		USPAT	2004/02/26 06:25
45		1		USPAT	2004/02/26 06:25
46		1	'	USPAT	2004/02/26 06:25
				USPAT	2004/02/26 06:25
47		. 1	·	i	
48		1		USPAT	2004/02/26 06:26
49		2	4704139.pn.	USPAT;	2004/02/26 06:34
17		_	1,01103.	US-PGPUB;	,,
		,		EPO; JPO;	· ·
			•	DERWENT	
50		111845	corrugated	USPAT;	2004/02/26 11:25
1 30		111043	Corrugated	'	2001,02,20 11.25
İ				US-PGPUB;	
ŀ				EPO; JPO;	
				DERWENT	
-1		,	/500/210 mala om 502/600 mala \ and		2004/02/26 07:20
51		1	(560/218.ccls. or 562/600.ccls.) and	USPAT;	2004/02/26 07:20
			corrugated	US-PGPUB;	· ·
	1		_	EPO; JPO;	
1				DERWENT	*
1		_			0004/00/05 07 05
52		2	6635148.pn.	USPAT;	2004/02/26 07:21
				US-PGPUB;	
				EPO; JPO;	
				DERWENT	
53		2	6436245.pn.	USPAT;	2004/02/26 07:22
			_	US-PGPUB;	1
				•	1
1			· ·	EPO; JPO;	1
			· · · · · · · · · · · · · · · · · · ·	DERWENT	1
54		1	6695928.pn.	USPAT;	2004/02/26 07:31
"			0000000.	,	,,,
1				US-PGPUB;	1
				EPO; JPO;	1
				DERWENT	1
		20000	nolymoriantion		2004/02/26 07:31
56		366209 <sub>.</sub>	polymerization	USPAT;	2004/02/26 07:31
				US-PGPUB;	1
				EPO; JPO;	1
1				DERWENT	
1					
57		2314	corrugated and polymerization	USPAT;	2004/02/26 07:32
1				US-PGPUB;	
1			,		
1			· '	EPO; JPO;	
1				DERWENT	
58		180	corrugated same polymerization	USPAT;	2004/02/26 07:32
1 30		100	corragated bame porymerraacton		=====================================
				US-PGPUB;	
1				EPO; JPO;	1
				DERWENT	

59	0	(560/218.ccls. or 562/600.ccls.) and (corrugated same polymerization)	USPAT; US-PGPUB; EPO; JPO;	2004/02/26 07:32
60	1	(560/218.ccls. or 562/600.ccls.) and corrugated	DERWENT USPAT; US-PGPUB;	2004/02/26 07:37
61	2	4865819.pn.	EPO; JPO; DERWENT USPAT;	2004/02/26 07:40
		100501319	US-PGPUB; EPO; JPO; DERWENT	
62	2	5296205.pn.	USPAT; US-PGPUB; EPO; JPO;	2004/02/26 12:54
63	7		DERWENT USPAT; US-PGPUB; EPO; JPO;	2004/02/26 07:43
64 65	1 0		DERWENT USPAT USPAT;	2004/02/26 07:43 2004/02/26 07:43
-			US-PGPUB; EPO; JPO; DERWENT	
67	3	"03127606"	USPAT; US-PGPUB; EPO; JPO; DERWENT	2004/02/26 07:53
68	9	RMax and corrugated	USPAT; US-PGPUB; EPO; JPO;	2004/02/26 08:20
69	555	203/60.ccls.	DERWENT USPAT; US-PGPUB;	2004/02/26 08:21
70	155	203/61.ccls.	EPO; JPO; DERWENT USPAT;	2004/02/26 08:21
			US-PGPUB; EPO; JPO; DERWENT	
71	635	203/60.ccls. or 203/61.ccls.	USPAT; US-PGPUB; EPO; JPO; DERWENT	2004/02/26 08:21
73	. 0	corrugated and ((560/218.ccls. or 562/600.ccls.) and (203/60.ccls. or 203/61.ccls.))	USPAT; US-PGPUB; EPO; JPO;	2004/02/26 08:22
72	32	(560/218.ccls. or 562/600.ccls.) and (203/60.ccls. or 203/61.ccls.)	DERWENT USPAT; US-PGPUB; EPO; JPO;	2004/02/26 08:25
74	167755	voids	DERWENT USPAT; US-PGPUB;	2004/02/26 08:26
75	301	210/696.ccls.	EPO; JPO; DERWENT USPAT;	2004/02/26 08:26
	06227		US-PGPUB; EPO; JPO; DERWENT	2004/02/25 08:32
80	262376	packing	USPAT; US-PGPUB; EPO; JPO;	2004/02/26 08:32
82	7	210/696.ccls. and packing	DERWENT USPAT; US-PGPUB; EPO; JPO;	2004/02/26 08:33
]		*	DERWENT	

83	185703	distillation	USPAT;	2004/02/26 09:16
			US-PGPUB;	
			EPO; JPO;	
		·	DERWENT	
0.4	20	210/696.ccls. and distillation	]	2004/02/26 08:44
84	20	210/696.ccis. and distillation	USPAT;	2004/02/20 08:44
			US-PGPUB;	
			EPO; JPO;	
			DERWENT	
85	2	5762668.pn.	USPAT;	2004/02/26 08:44
		•	US-PGPUB;	
			EPO; JPO;	
1			DERWENT	
0.0	366	(560/218.ccls. or 562/600.ccls.) and	USPAT;	2004/02/26 08:48
86	300	distillation		2004/02/20 00:40
		distillation	US-PGPUB;	
		, ·	EPO; JPO;	•
		•	DERWENT	
87	1368	packing adj layer	USPAT;	2004/02/26 08:48
		• ,	US-PGPUB;	
			EPO; JPO;	
			DERWENT	
88	0	((560/218.ccls. or 562/600.ccls.) and	USPAT;	2004/02/26 08:49
00		distillation) and (packing adj layer)	US-PGPUB;	
		distillation, and (packing adj layer)	EPO; JPO;	
			1	
			DERWENT	
89	262376	packing	USPAT;	2004/02/26 09:15
		*	US-PGPUB;	
	0.		EPO; JPO;	
			DERWENT	
90	85	((560/218.ccls. or 562/600.ccls.) and	USPAT;	2004/02/26 08:50
,		distillation) and packing	US-PGPUB;	
*		distillation, and packing	EPO; JPO;	
		·		
	-		DERWENT	
91	62	RMax and packing	USPAT	2004/02/26 09:13
93	O.	RMax same distillation	USPAT;	2004/02/26 09:16
	,	•	US-PGPUB;	
1		•2	EPO; JPO;	
			DERWENT	
94	18	RMax and distillation	USPAT;	2004/02/26 09:22
74	10	Max and distillation	US-PGPUB;	2001, 02, 20
			EPO; JPO;	
				]
		·	DERWENT	
95	13047	Ry	USPAT;	2004/02/26 09:22
		·	US-PGPUB;	
			EPO; JPO;	
			DERWENT	l i
96	1011	Ry and distillation	USPAT;	2004/02/26 09:23
		1	US-PGPUB;	
			EPO; JPO;	1
			DERWENT	
		D distillation		2004/02/26 00-22
97	10	Ry same distillation .	USPAT;	2004/02/26 09:23
			US-PGPUB;	
		Ye	EPO; JPO;	
			DERWENT	1
98	7897	corrugated and packing	USPAT;	2004/02/26 11:25
1			US-PGPUB;	1
			EPO; JPO;	
			DERWENT	
	0044335	-1		2004/02/26 11:25
99	2944323	plate	USPAT;	2004/02/20 II:25
			US-PGPUB;	1
			EPO; JPO;	
			DERWENT	
100	9546	corrugated near3 plate	USPAT;	2004/02/26 11:26
			US-PGPUB;	
			EPO; JPO;	
			DERWENT	
	-)-	/		2004/02/26 11-26
101	635	(corrugated near3 plate) and packing	USPAT;	2004/02/26 11:26
			US-PGPUB;	
			EPO; JPO;	
			DERWENT	
	1	1		·

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103	6	(((corrugated near3 plate) and packing) and	USPAT;	2004/02/26 11:27
		distillation) and polymerization	US-PGPUB;	
		·	EPO; JPO;	
			DERWENT	
102	110	, , ,	USPAT;	2004/02/26 11:29
		distillation	US-PGPUB;	
			EPO; JPO;	
1			DERWENT	0004/00/05 00
104	98183	polymerizable	USPAT;	2004/02/26 11:30
			US-PGPUB;	
		·	EPO; JPO; DERWENT	-
105	2	(((corrugated near3 plate) and packing) and	USPAT;	2004/02/26 11:30
103		distillation) and polymerizable	US-PGPUB:	2004/02/20 11:30
		distillation, and polymerizable	EPO; JPO;	·
		,	DERWENT	
106	18	corregated adj plate	USPAT;	2004/02/26 12:55
		, , , , , , , , , , , , , , , , , , ,	US-PGPUB:	
			EPO; JPO;	i i
			DERWENT	
107	3	"gas-injection" near3 plate	USPAT;	2004/02/26 12:57
			US-PGPUB;	*
[ ,			EPO; JPO;	
			DERWENT	

	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error Definition
				·	USPAT		·	
1	BRS	L9	264	560/218.ccls.	US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:48		
2	BRS	L10	388	562/600.ccls.		2004/02/26 05:48		
3	BRS	L11	0	5296205.URPN.	USPAT	2004/02/26 05:48		
4	BRS	L12	0	5296205.URPN.	USPAT	2004/02/26 05:48		
5	BRS	L13	475	261/97.ccls.		2004/02/26 05:48		
6	BRS	L14	627	560/218.ccls. or 562/600.ccls.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:47		
7	BRS	L15	319	"gas-injection"		2004/02/26 12:56		
8	BRS	<b>L</b> 17	6821	"beam type"	EPO; JPO; DERWE NT	2004/02/26 05:49		
9	BRS	L18	0	(560/218.ccls. or 562/600.ccls.) and "gas-injection"	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:49		

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	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error	Definition
10	BRS	L19	0 .	(560/218.ccls. or 562/600.ccls.) and "beam type"	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:49			·
11	BRS	L20	0	261/97.ccls. and "gas-injection"	EPO; JPO; DERWE NT	2004/02/26 05:49			
12	BRS	L21	10583 4	support adj plate	EPO; JPO; DERWE NT	2004/02/26 05:49			
13	BRS	L22	242	"beam type" and (support adj plate)	EPO; JPO; DERWE NT	2004/02/26 05:49			
14	BRS	L23	0	"beam type" near4 (support adj plate)	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:49			
15	BRS	L27	1737	RMax	USPAT	2004/02/26 09:13			
16	BRS	L28	0	("gas-injection" near4 (support adj plate)) and RMax	USPAT; ; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:49			-
17	BRS	L1	2	US-5296205-\$.DID.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 05:49			-

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	Туре	L #	Hits	Search Text	DBs	Time St	tamp	Comments	Error	Definition
18	BRS	L30	0	261/97.ccls. and RMax		2004/02 08:49	2/26			
19	BRS	L2	1	"4011304".PN.	USPAT	2004/02 05:49	2/26			
20	BRS	L3	1	"4108682".PN.	USPAT	2004/02 05:49	2/26			
21	BRS	L4	1	"4125597".PN.	USPAT	2004/02 05:49	2/26			
22	BRS	L5	1	"4189462".PN.	USPAT	2004/02 05:49	2/26			
23	BRS	L6	1	"4455287".PN.	USPAT	2004/02 05:49				
24	BRS	L7	1	"4668405".PN.	USPAT	2004/02 05:49				
25	BRS	L8	1	"4719020".PN.	USPAT	2004/02 05:49	2/26			
26	BRS	L16	2	261/97.ccls. and (560/218.ccls. or 562/600.ccls.)		2004/02 05:49	2/26			
27	BRS	L24	12	"beam type" same (support adj plate)	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02 06:20	2/26			
28	BRS	L25	2	"gas-injection" near4 (support adj plate)	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02 05:49	×			
29	BRS	L26	4	4814117.URPN.	USPAT	2004/02 05:49	2/26			
30	BRS	L29	1	(560/218.ccls. or 562/600.ccls.) and RMax	USPAT	2004/02 05:49	2/26			
31	BRS	L31	1	"2098667".PN.	USPAT	2004/02 05:54				
32	BRS	L32	1	"2158229".PN.	USPAT	2004/02 05:54				
33	BRS	L33	1	"2283307".PN.	USPAT	05:54				
34	BRS	L34	1	"2987039".PN.	USPAT	05:54				
35	BRS	L35	1	"3653845".PN.	USPAT	2004/0: 05:55	2/26			

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	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error Definition
36	BRS	L36	1	"4003178".PN.	USPAT	2004/02/26 05:55		
37	BRS	L37	1	"4275018".PN.	USPAT	2004/02/26 05:55		
38	BRS	L38	1	"4452025".PN.	USPAT	2004/02/26 05:56	·	
39	BRS	L39	1	"4472325".PN.	USPAT	2004/02/26 05:56		_
40	BRS	L40	1	"4557876".PN.	USPAT	2004/02/26 05:57		×
41	BRS	L41	8	115 and 121		2004/02/26 06:32		
42	BRS	L42	5	4576763.URPN.	USPAT	2004/02/26 06:23	·	
43	BRS	L43	1	"2212932".PN.	USPAT	2004/02/26 06:25		
44	BRS	L44	1	"2602651".PN.	USPAT	2004/02/26 06:25		
45	BRS	L45	1	"3266787".PN.	USPAT	2004/02/26 06:25		
46	BRS	L46	1	"3266787".PN.	USPAT	2004/02/26 06:25		
47	BRS	L47	1.	"4041113".PN.	USPAT	2004/02/26 06:25		
48	BRS	L48	1	"4067936".PN.	USPAT	2004/02/26 06:26		
49	BRS	L49	2	4704139.pn.		2004/02/26 06:34		
50	BRS	L50	11184 5	corrugated	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 11:25		
51	BRS	L51	1	114 and 150	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 07:20		

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	Туре	L #	Hits	Search Text	DBs	Time	Stamp	Comments	Error	Definition
52	BRS		2	6635148.pn.	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/	02/26			
53	BRS	L53	2	6436245.pn.	EPO; JPO; DERWE NT	2004/ 07:22				
54	BRS	L54	1	6695928.pn.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:31				
55	BRS	L56	36620 9	polymerization	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:31				
56	BRS	L57	2314	150 and 156	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/			•	
57	BRS	L58	180	150 same 156	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:32				
58	BRS	L59	0	114 and 158	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:32	02/26			

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53.	0
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57	0
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	Туре	L #	Hits	Search Text	DBs	Time	Stamp	Comments	Error	Definitio
59	BRS	L60	1	114 and 150		2004/ 07:37				
60	BRS	L61	2	4865819.pn.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:40	02/26			
61	BRS	L62	2	5296205.pn.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 12:54				
62	BRS	L63	7	12 or 13 or 14 or 15 or 16 or 17 or 18	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 07:43				
63	BRS	L64	1	"4622212".PN.	USPAT	2004/ 07:43	02/26			
64	BRS	L65	0	114 and 163	USPAT; ; US-PG PUB; EPO; JPO; DERWE NT		02/26			
65	BRS	<b>L</b> 67	3	"03127606"	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/ 07:53	(02/26 3			
66	BRS	L68	9	127 and 150	USPAT; US-PG PUB; EPO; JPO; DERWE	2004, 08:20	(02/26 )			

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59	0
60	0
61	0
62	0
63	0
64	0
65	0
66	0

	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error	Definition
67	BRS	L69	555	203/60.ccls.	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/02/26 08:21			
68	BRS	L70	155	203/61.ccls.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:21			
69	BRS	L71	635	169 or 170	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:21			
70	BRS	L73	0	150 and 172	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:22			•
71	BRS	L72	32	l14 and 171	USPAT; ; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:25			
72	BRS	L74	16775 5	voids	USPAT; ; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:26			
73	BRS	L75	301	210/696.ccls.	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/02/26 08:26			

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67	0
68	0
69	0
 70	0
71	0
72	0
73	0

	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error Definition
74	BRS	L80	26237 6	packing	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/02/26 08:32	-	
75	BRS	L82	7	175 and 180	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:33		
	BRS	L83	18570 3	distillation	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 09:16		
77	BRS	L84	20	175 and 183	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:44		
78	BRS	L85	2	5762668.pn.	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:44		
79	BRS	L86	366	l14 and 183	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:48		
80	BRS	L87	1368	packing adj layer	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 08:48		

74 0 75 0 76 0 77 0 78 0 79 0		Err ors
76 0 77 0 78 0	74	0
77 O 78 O 79 O	75	0
78 0	76	0
79 0	77	0
	78	0
80 0	79	0
	80	0

	Туре	L #	Hits	Search Text	DBs	Time	Stamp	Comments	Error	Definition
81	BRS	L88	0	186 and 187		2004/ 08:49	02/26			
82	BRS	L89	26237 6	packing	EPO; JPO; DERWE NT	2004/ 09:15	02/26			
83	BRS	L90	85	186 and 189	USPAT; US-PG PUB; EPO; JPO; DERWE NT	08:50	,			
84	BRS	L91	62	127 and 180	USPAT	2004/ 09:13	02/26			
85	BRS	L93	0	127 same 183	USPAT; US-PG PUB; EPO; JPO; DERWE NT		02/26			
86	BRS	L94	18	127 and 183	USPAT; US-PG PUB; EPO; JPO; DERWE NT		(02/26 !			
87	BRS	L95	13047	Ry	EPO; JPO; DERWE NT		′02/26 ?			-
88	BRS	L96	1011	195 and 183	USPAT; US-PG PUB; EPO; JPO; DERWE NT	09:23	/02/26 }			

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81	ors 0
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	Туре	L #	Hits	Search Text	DBs	Time Stamp	Comments	Error	Definition
89	BRS	L97	10	195 same 183	EPO; JPO; DERWE NT	2004/02/26 09:23			
90	BRS	L98	7897	150 and 189	EPO; JPO; DERWE NT	2004/02/26 11:25			
91	BRS	L99	29443 23	plate	EPO; JPO; DERWE NT	2004/02/26 11:25			
92	BRS	L100	9546	150 near3 199	USPAT; US-PG PUB; EPO; JPO; DERWE NT				
93	BRS	L101	635	l100 and 189	USPAT ; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 11:26			
94	BRS	L103	6	l102 and 156	USPAT ; US-PG PUB; EPO; JPO; DERWE NT	2004/02/26 11:27	8		
95	BRS	L102	110	l101 and 183	USPAT; US-PG PUB; EPO; JPO; DERWE	2004/02/26 11:29			

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89	0
90	0
91	0
92	0
93	0
94	0
95	0

	Туре	L #	Hits	Search Text	DBs	Time	Stamp	Comments	Error	Definition
96	BRS	L104	98183	polymerizable		2004/ 11:30	02/26			
97	BRS	L105	2	1102 and 1104	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 11:30				
98	BRS	L106	18	corregated adj plate	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 12:55	02/26			
99	BRS	L107	3	l15 near3 l99	USPAT; US-PG PUB; EPO; JPO; DERWE NT	2004/ 12:57				

	Err
96	0
97	0 0
98	0
99	0

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NEWS
     2
                 "Ask CAS" for self-help around the clock
                 CA/CAplus records now contain indexing from 1907 to the
NEWS
        SEP 09
                 present
NEWS
        DEC 08
                 INPADOC: Legal Status data reloaded
NEWS
        SEP 29
                 DISSABS now available on STN
        OCT 10
                PCTFULL: Two new display fields added
NEWS 6
        OCT 21
                 BIOSIS file reloaded and enhanced
NEWS 7
NEWS 8 OCT 28
                 BIOSIS file segment of TOXCENTER reloaded and enhanced
        NOV 24
                MSDS-CCOHS file reloaded
NEWS 9
NEWS 10
        DEC 08
                CABA reloaded with left truncation
        DEC 08
NEWS 11
                 IMS file names changed
                 Experimental property data collected by CAS now available
NEWS 12
       DEC 09
                 in REGISTRY
                 STN Entry Date available for display in REGISTRY and CA/CAplus
NEWS 13
        DEC 09
NEWS 14
        DEC 17
                 DGENE: Two new display fields added
NEWS 15
        DEC 18
                 BIOTECHNO no longer updated
                 CROPU no longer updated; subscriber discount no longer
NEWS 16 DEC 19
                 available
                Additional INPI reactions and pre-1907 documents added to CAS
NEWS 17
        DEC 22
                 databases
        DEC 22
                 IFIPAT/IFIUDB/IFICDB reloaded with new data and search fields
NEWS 18
                ABI-INFORM now available on STN
NEWS 19
        DEC 22
NEWS 20
                 Source of Registration (SR) information in REGISTRY updated
        JAN 27
                 and searchable
                 A new search aid, the Company Name Thesaurus, available in
NEWS 21
        JAN 27
                 CA/CAplus
                 German (DE) application and patent publication number format
NEWS 22
        FEB 05
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              AND CURRENT DISCOVER FILE IS DATED 23 SEPTEMBER 2003
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              CAS World Wide Web Site (general information)
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=> ile caplus
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index. Enter "HELP COMMANDS" at an arrow prompt (=>) for a list of

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FILE COVERS 1907 - 26 Feb 2004 VOL 140 ISS 9 FILE LAST UPDATED: 25 Feb 2004 (20040225/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> gas injection 1343505 GAS 464127 GASES 1510424 GAS

(GAS OR GASES)

428627 INJECTION 97499 INJECTIONS 488252 INJECTION

(INJECTION OR INJECTIONS)

3818 GAS INJECTION

(GAS (W) INJECTION)

=> acryl?

L2 402843 ACRYL?

=> 11 and 12

L3 20 L1 AND L2

=> d 12 10-20 ti

L2 ANSWER 10 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN

TI Improvement of acrylonitrile reactor control system

-)

- L2 ANSWER 11 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Real-Time Infrared Determination of Photoinitiated Copolymerization Reactivity Ratios: Application of the Hilbert Transform and Critical Evaluation of Data Analysis Techniques
- L2 ANSWER 12 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Self-crosslinking acrylic elastic emulsions and aggregates and putties therefrom
- L2 ANSWER 13 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polyolefin-based filler and its preparation
- L2 ANSWER 14 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Screening of Catalysts for Acrylonitrile Decomposition
- L2 ANSWER 15 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis and characterization of poly (aniline-co-acrylonitrile ) using organic benzoyl peroxide by inverted emulsion method
- L2 ANSWER 16 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Preparation of halo-tolerant, high absorbent resin
- L2 ANSWER 17 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI In Situ Time-Dependent Signatures of Light Scattered from Solutions undergoing Polymerization Reactions
- L2 ANSWER 18 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polyelectrolyte Complexes of Chitosan and Poly(acrylic acid) As Proton Exchange Membranes for Fuel Cells
- L2 ANSWER 19 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Determination of mole ratio of monomers in new adsorbent of solid-phase microextraction by infrared spectroscopy
- L2 ANSWER 20 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis of the 3-cyanethyl-1,3-oxazacycloalkanes. New of data about the existence of ring-chain tautomerism in the row of  $\beta\text{-or}\gamma\text{-}$  hydroxylcontaining imines
- => polym?
- L4 1868920 POLYM?
- => d 12 1-9 ti
- L2 ANSWER 1 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Measurement of the dissolved oxygen concentration in acrylate monomers with a novel photochemical method
- L2 ANSWER 2 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Fluorescent probes for monitoring the pulsed-laser-induced photocuring of poly(urethane acrylate)-based adhesives
- L2 ANSWER 3 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Silicone-based impact modifiers for poly(vinyl chloride), engineering resins, and blends
- L2 ANSWER 4 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Emulsion polymerization: From fundamental mechanisms to process developments
- L2 ANSWER 5 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Phase separation of off-critical polymer blends of poly(styrene-co-maleic

anhydride) and poly(methyl methacrylate). II. Morphology and mechanical properties

- L2 ANSWER 6 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Flammability of butadiene-acrylonitrile rubbers
- L2 ANSWER 7 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Conformational alteration of bradykinin in presence of GM1 micelle
- L2 ANSWER 8 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Non-sequencing molecular approaches to identify preS2-defective hepatitis B virus variants proved to be associated with severe liver diseases
- L2 ANSWER 9 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis and application of 3-picoline

### => d 12 4 ti fbib abs

- L2 ANSWER 4 OF 402843 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Emulsion polymerization: From fundamental mechanisms to process developments
- AN 2004:152353 CAPLUS
- TI Emulsion polymerization: From fundamental mechanisms to process developments
- AU Asua, Jose M.
- CS Institute for Polymer Materials (POLYMAT) and Grupo de Ingenieria Quimica, Facultad de Ciencias Quimicas, The University of the Basque Country, Donostia-San Sebastian, 20018, Spain
- Journal of Polymer Science, Part A: Polymer Chemistry (2004), 42(5), 1025-1041 CODEN: JPACEC; ISSN: 0887-624X
  - John Wiley & Sons, Inc.
- PB John Wiley & DT Journal
- LA English
- AB Emulsion polymers are "products by process" whose main properties are determined during polymerization. In this scenario of margins reduction, increasing

competition, and public sensitivity to environmental issues, the challenge is to achieve an efficient production of high-quality materials in a consistent, safe, and environmentally friendly way. This highlight reviews the investigations carried out at The University of the Basque Country to develop a knowledge-based strategy to achieve these goals. First, the research in fundamental mechanisms is discussed. This includes studies in radical entry and exit, oil-soluble initiators, propagation accounts. of acrylic monomers, processes involved in the formation of branched and crosslinked polymers, microstructure modification by postreaction operations, the formation of particle morphol., and reactive surfactants. The advanced math. models developed in the group are also reviewed. In the second part, the advances in process development (optimization, online monitoring and control, monomer removal, production of high-solids, low-viscosity latices, and process intensification) are presented.

## => d his

L1

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004

- 3818 GAS INJECTION
- L2 402843 ACRYL?
- L3 20 L1 AND L2

- => 11 and 14
- L5 159 L1 AND L4
- => packing

69768 PACKING

6972 PACKINGS

L6 72711 PACKING

(PACKING OR PACKINGS)

- => 11 and 16
- L7 18 L1 AND L6
- => d 17 1-18 ti
- L7 ANSWER 1 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Liquid holdup in non-wetting packing with lateral Gas injection
- L7 ANSWER 2 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI External gas moulding
- L7 ANSWER 3 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Process and apparatus for injection moulding hollow plastic articles
- L7 ANSWER 4 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI A 3-D finite element model for gas-assisted injection molding: simulations and experiments
- L7 ANSWER 5 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Foam flow in heterogeneous porous media: effect of crossflow
- L7 ANSWER 6 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Computer simulation and experimental verification of gas-assisted injection molding
- L7 ANSWER 7 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI On the dynamics of gas-assisted injection molding process
- L7 ANSWER 8 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI An additive for increasing the density of a fluid and fluid comprising such additive
- L7 ANSWER 9 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI On the dynamics of gas-assisted injection molding process
- L7 ANSWER 10 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Filling and packing CAE software for gas injection molding
- L7 ANSWER 11 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Natural gas storage Rehden (Germany). Planning, construction, and start-up
- L7 ANSWER 12 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Performance of air current distribution in shallow bed **packing** columns with side inlets
- L7 ANSWER 13 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Permeable ceramics utilizing spherical particles
- L7 ANSWER 14 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Permeable refractories using spherical particles

```
ANSWER 15 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
L7
    High-intensity gas/liquid mass transfer in the bubbly flow region during
TI
    co-current upflow through static mixers
    ANSWER 16 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
T.7
    Separation of oil-water mixtures in particulate beds
TΙ
    ANSWER 17 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN
L7
    Distillation tower for petroleum and its derivaties
```

ANSWER 18 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN L7

Gas-liquid chromatography of lipids TТ

# => d 17 15 ti fbib abs

ANSWER 15 OF 18 CAPLUS COPYRIGHT 2004 ACS on STN **T.7** High-intensity gas/liquid mass transfer in the bubbly flow region during TI

1985:97506 CAPLUS AN

DИ 102:97506

ΤI

High-intensity gas/liquid mass transfer in the bubbly flow region during TT co-current upflow through static mixers

Roes, A. W. M.; Zeeman, A. J.; Bukkems, F. H. J. ΑIJ Koninklijke/Shell-Lab., Amsterdam, 1003 AA, Neth. CS

co-current upflow through static mixers

Institution of Chemical Engineers Symposium Series (1984), 87 (Chem. React. SO Eng.), 231-8 CODEN: ICESDB; ISSN: 0307-0492

DTJournal

LΑ English

=> distill?

The bubble-slug flow transition, pressure drop, gas hold-up, bubble diameter, AΒ mass-transfer coeffs., and power input were examined with air-water system in static mixers. A liquid circulation loop was used with gas injection and 3 packings: a com. static mixer, Raschig rings, and wire gauze. The liquid mass-transfer coefficient was correlated to the pump-power input. A correlation of the bubble-slug flow transition is qiven.

# 108582 DISTILL? 156850 DISTD 1 DISTDS 156850 DISTD (DISTD OR DISTDS) 23949 DISTG 167594 DISTN

1717 DISTNS 168313 DISTN

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rs353964 DISTILL?

(DISTILL? OR DISTD OR DISTG OR DISTN)

### $\cdot => d his$

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004

3818 GAS INJECTION L1

L2 402843 ACRYL?

L3 20 L1 AND L2

1868920 POLYM?

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L5
           159 L1 AND L4
L6
         72711 PACKING
            18 L1 AND L6
L7
        353964 DISTILL?
1.8
=> 11 \text{ and } 18
           38 L1 AND L8
=> 14 and 19
            1 L4 AND L9
L10
=> d l10 ti fbib bas
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OIBIB ----- OBIB, indented with text labels
SBIB ----- BIB, no citations
SIBIB ----- IBIB, no citations
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HITIND ----- IC, ICA, ICI, NCL, CC and index field (ST and IT)
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HITSEQ ----- HIT RN, its text modification, its CA index name, its
             structure diagram, plus NTE and SEQ fields
FHITSTR ---- First HIT RN, its text modification, its CA index name, and
             its structure diagram
FHITSEQ ---- First HIT RN, its text modification, its CA index name, its
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## => d l10 ti fbib abs

L10 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2004 ACS on STN

TI Synthetic lubricants by oligomerization and hydrogenation

AN 1974:98214 CAPLUS

DN 80:98214

TI Synthetic lubricants by oligomerization and hydrogenation

IN Shubkin, Ronald L.

PA Ethyl Corp.

SO U.S., 5 pp. CODEN: USXXAM

DT Patent

LA English

FAN. CNT 1

PATENT NO.		KIND DATE		APPLICATION NO.	DATE	
ΡI	US 3780128	Α	19731218	US 1971-195443 US 1971-195443	19711103 19711103	

AB Synthetic lubricating fluids, of low pour point and high viscosity index, are prepared by polymerizing C6-16 α-olefins, using alc.-promoted BF3 in molar excess of the alc. The fluids are improved by hydrogenation. Thus, 1-decene and a preformed BF3-n-propanol complex were mixed 10 min. at room temperature with no reaction. Injection of BF3 gas initiated a reaction and temperature was held at 27° for 10 min. when anal. gave monomer 10, dimer 10 and trimer 80%. Further gas injection for 20 min. caused a reaction with temperature reaching 35° and in 1 hr almost complete polymerization had occurred. This polymer was hydrogenated on Ni-containing kieselguhr catalyst at 220° and 900-1000 psi for 10 hr. After filtering, the oil was stripped of monomer and distilled in vacuo to remove the dimer. This gave an oil of 22.72 at 100 and 4.6 cSt at 210°F with a viscosity index 131 and a pour point -80°F.

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CA SUBSCRIBER PRICE	ENTRY -2.08	SESSION -2.08
	ENTRY -2.08	SESSION -2.08

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=> d his

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004 3818 GAS INJECTION L1402843 ACRYL? L220 L1 AND L2  $^{\text{L3}}$ 1868920 POLYM? L4159 L1 AND L4 L5 72711 PACKING L6 18 L1 AND L6 ь7 353964 DISTILL?  $r_8$ 38 L1 AND L8 L91 L4 AND L9 L10

=> corrug?

L11 15009 CORRUG?

=> 11 and 111

L12 2 L1 AND L11

=> d 112 1-2 ti

- L12 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Land disposal of waste gases. III. Sorption patterns from buried gas injection pipes
- L12 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Land disposal of waste gases. II. Gas flow from buried pipes

=> 16 and 111

L13 542 L6 AND L11

=> 14 and 113

L14 44 L4 AND L13

=> d 114 24-44 ti

- L14 ANSWER 24 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Manufacture of porous carbon plate
- L14 ANSWER 25 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packings for wastewater treatment

- L14 ANSWER 26 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI An ordered bed packing module
- L14 ANSWER 27 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Molecular compounds and complexes. Part XII. Crystal chemistry of N-(p-tolyl)tetrachlorophthalimide and of its channel inclusion complexes
- L14 ANSWER 28 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Crazing in two polystyrene/polybutadiene block copolymers
- L14 ANSWER 29 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Shock-absorbing packing materials
- L14 ANSWER 30 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Corrugated paperboard of the monowave or multiwave type for packing
- L14 ANSWER 31 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Transmission electron diffraction intensities from real organic crystals: thin plate microcrystals of paraffinic compounds
- L14 ANSWER 32 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packing for effecting contact of a gas with a flowing liquid film
- L14 ANSWER 33 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Complex products based on foam and a porous reinforcement agent
- L14 ANSWER 34 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Testing of combinations made of paper and polystyrene foam film
- L14 ANSWER 35 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Size for corrugated cardboard
- L14 ANSWER 36 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Sintered plates of poly(vinyl chloride) resin for use as **packing** in cooling towers
- L14 ANSWER 37 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Apparatus for continuous polycondensation reaction
- L14 ANSWER 38 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Heat-sealable packing material
- L14 ANSWER 39 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packing for gas-liquid contact systems
- L14 ANSWER 40 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Some new habit features in crystals of long chain compounds. IV. The fold surface geometry of monolayer polyethylene crystals and its relevance to fold packing and crystal growth
- L14 ANSWER 41 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Some new habit features in crystals of long-chain compounds. II. Polymers
- L14 ANSWER 42 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Molded polystyrene for trickling filters
- L14 ANSWER 43 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Corrosion-inhibiting coating for separator and packing cartons

- L14 ANSWER 44 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI American Society for Testing Materials, Standards, 1943. Supplement. III. Nonmetallic materials, general

# => d 114 39 ti fbib abs

L14 ANSWER 39 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN

FI Packing for gas-liquid contact systems

AN 1967:86414 CAPLUS

DN 66:86414

TI Packing for gas-liquid contact systems

IN Moeller, Delmer H.; Towsley, Frank E.; Moss, Charles F.

PA Dow Chemical Co.

SO U.S., 5 pp. CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE .	APPLICATION	NO.	DATE	•	•
PI	US 3281307		19661025	US		19621105		
AB	Extended surface							
	oxidation towers							
	corrugated sheet	s of a	molded rigid t	hermoplastic	mate	rial, e.g.	,	
	polystyrene, unp	lastic	ized poly(vinyl	chloride),	saran	, acrylic,	or a	
	thermosetting ph							
	corrugated on tw	o port	ions separated	by a sinusoi	dal t	ransition		
	section. To pre							
	of the 2 portion	s are	displaced by .a	pprx.180° and	d the	sheets ar	e	
	joined by heat-s	ealing	, mech. fastnes	s, adhesives	, or	flange-wel	ding (U.S.	
	3,007,834). The	geome	tric configurat	ion of the s	heets	contribut	es high	
	compressive stre							
	from top to bott	om of	a packed tower,	simplifies	manuf	acture and	assembly,	and

=> logoff hold		
COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	49.98	50.19
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-2.77	-2.77

SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 07:00:39 ON 26 FEB 2004

allows nesting for space-saving during shipment.

Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID:SSSPTA1623PAZ

### PASSWORD:

\* \* \* \* \* \* RECONNECTED TO STN INTERNATIONAL \* \* \* \* \* \* SESSION RESUMED IN FILE 'CAPLUS' AT 07:10:54 ON 26 FEB 2004 FILE 'CAPLUS' ENTERED AT 07:10:54 ON 26 FEB 2004 COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	49.98	50.19
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	-2.77	-2.77

=> d his

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

```
FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004
           3818 GAS INJECTION
T.1
L2
         402843 ACRYL?
             20 L1 AND L2
L3
        1868920 POLYM?
L4
            159 L1 AND L4
L5
          72711 PACKING
L6
L7
             18 L1 AND L6
         353964 DISTILL?
L8
Ь9
             38 L1 AND L8
              1 L4 AND L9
L10
          15009 CORRUG?
L11
              2 L1 AND L11
L12
             542 L6 AND L11
L13
              44 L4 AND L13
L14
```

# => d 114 1-23 ti

- L14 ANSWER 1 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Insulating packaging material and related packaging system
- L14 ANSWER 2 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Solid-state morphologies of linear and bottlebrush-shaped polystyrene-poly(Z-L-lysine) block copolymers
- L14 ANSWER 3 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packing for mass-transfer and separation equipment
- L14 ANSWER 4 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Lithography printing plate **packing** structure, lithography printing plate packaging method and packaging material for lithography printingplate. [Machine Translation].
- L14 ANSWER 5 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Pattern formation and morphology control of the corrugation on the surface of photocurable polymer
- L14 ANSWER 6 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Structure and Optical Properties of Several Organic-Inorganic Hybrids Containing Corner-Sharing Chains of Bismuth Iodide Octahedra
- L14 ANSWER 7 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Suction of water sheet for foliage plant vegetable. [Machine Translation].
- L14 ANSWER 8 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Structures of ionic di(arenesulfonyl)amides. Part 2. Silver(I) di(arenesulfonyl)amides and a silver(I) (arenesulfonyl)(alkanesulfonyl)ami de. From ribbons to lamellar layers exhibiting short C-H···Hal-C or C-Br···Br-C interlayer contacts

- L14 ANSWER 9 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Solid-state supramolecular chemistry of porphyrins. Ligand-bridged tetraphenylmetalloporphyrin dimers
- L14 ANSWER 10 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Sensitive adhesive and its utilization
- L14 ANSWER 11 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for sealing solid **polymer** electrolyte fuel cells
- L14 ANSWER 12 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI A general boundary condition for liquid flow at solid surfaces
- L14 ANSWER 13 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis of MTBE. Influence of crosslinking and **polymer** content on activity and selectivity of **polymer** carrier ion exchange catalysis
- L14 ANSWER 14 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Separation of high concentration divinylbenzene
- L14 ANSWER 15 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for sealing solid polymer electrolyte fuel cells
- L14 ANSWER 16 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Influence of crosslinking and **polymer** content on activity and selectivity of **polymer**/carrier ion exchange catalysts for the synthesis of MTBE
- L14 ANSWER 17 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Cushioning materials from soft foamed polyurethanes
- L14 ANSWER 18 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI A novel nickel-containing chain-like **polymer:** [{Ni(NH3)4}Ni{S2C=C(CN)2}2]∞
- L14 ANSWER 19 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI High-temperature braided **packing** comprising a core of folded flexible graphite tape
- L14 ANSWER 20 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Manufacture of **packing** materials for mixing, mass-transfer, or heat-exchange apparatus
- L14 ANSWER 21 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Crystal structures and **polymorphism** in aliphatic p-amidobenzoic acids
- L14 ANSWER 22 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Plastic film composites as thermal insulators for containers
- L14 ANSWER 23 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packing material which releases sulfur dioxide in the presence of moisture
- => d 114 3,20 ti fbib abs
- L14 ANSWER 3 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packing for mass-transfer and separation equipment
- AN 2002:36726 CAPLUS
- DN 136:71611

```
TI Packing for mass-transfer and separation equipment
```

IN Vybornov, V. G.

PA Russia

SO Russ., No pp. given

CODEN: RUXXE7

DT Patent

LA Russian

FAN.CNT 1

P	PATENT NO.		DATE	APPLICATION NO.	DATE
PI R	u 2155095	C1	20000827		19990309 19990309

The packing has retainers made in the form of parallel rods AΒ which are mounted in a grid and between grid layers so that they partially overlap the packing cross section. Ends of the rods are fastened by means of plates. The grid layers are positioned at an angle to the rods. The rods may be positioned at an angle to a horizontal plane to facilitate draining of liqs. The grid is made in the form of a multilayer packet and may have a zigzag-shaped or arch-shaped profile along the height or length. The grid may be manufactured from a woven or welded mesh and may have cells of different dimensions and shapes, or may be manufactured from wire or monofilament of a different thickness and profile. A sieve sleeve woven from a wire or polymer monofilament may be used as the grid. The sleeve may be flat or preliminarily corrugated. The packing has an increased efficiency in mass exchange and separation processes, wider operational capabilities, and simplified construction.

L14 ANSWER 20 OF 44 CAPLUS COPYRIGHT 2004 ACS on STN

TI Manufacture of **packing** materials for mixing, mass-transfer, or heat-exchange apparatus

AN 1993:674345 CAPLUS

DN 119:274345

TI Manufacture of **packing** materials for mixing, mass-transfer, or heat-exchange apparatus

IN Nagaoka, Tadayoshi

PA Nagaoka Intl. Corp., Japan

SO Faming Zhuanli Shenqing Gongkai Shuomingshu, 23 pp. CODEN: CNXXEV

DT Patent

LA Chinese

FAN.CNT 1

				3 5 5 7 7 6 3 M T 6 1/ 1/ 6	DAME
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	CN 1070845	A	19930414	CN 1992-111344	19920930
				JP 1991-280548 A	19911001
	JP 05096101	A2	19930420	JP 1991-280548	19911001
	JP 3252972	B2	20020204		
	JP 2002201550	A2	20020719	JP 2001-317940	19911001
				JP 1991-280548 A	319911001

AB The packing materials are manufactured by knitting parallel corrugated permeable sheets to form multiple channels in main flow direction for flowing fluids and protrusions of the sheets in transverse direction of the main flow, and simultaneously joining the protrusions of adjacent sheets by knitting. The packing materials may be made of metal wires, synthetic, plant, and/or glass fibers.

=> logoff hold
COST IN U.S. DOLLARS
SINCE FILE TOTAL
ENTRY SESSION
FULL ESTIMATED COST
67.12
67.33

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE TOTAL
ENTRY SESSION

CA SUBSCRIBER PRICE

-4.16
-4.16

SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 07:17:34 ON 26 FEB 2004

Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID:SSSPTA1623PAZ

## PASSWORD:

\* \* \* \* \* RECONNECTED TO STN INTERNATIONAL \* \* \* \* \* \* SESSION RESUMED IN FILE 'CAPLUS' AT 07:54:15 ON 26 FEB 2004 FILE 'CAPLUS' ENTERED AT 07:54:15 ON 26 FEB 2004 COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

COST IN U.S. DOLLARS

SINCE FILE TOTAL
ENTRY SESSION
FULL ESTIMATED COST

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

CA SUBSCRIBER PRICE

SINCE FILE TOTAL
ENTRY SESSION
-4.16
-4.16

=> d his

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004 3818 GAS INJECTION , L1402843 ACRYL? L220 L1 AND L2 L3 1868920 POLYM? L4159 L1 AND L4 L5 L6 72711 PACKING 18 L1 AND L6 L7353964 DISTILL?  $\Gamma8$ L9 38 L1 AND L8 L101 L4 AND L9 L11 15009 CORRUG? L122 L1 AND L11 542 L6 AND L11 L13 44 L4 AND L13 L14

=> 14 and 16

L15 10957 L4 AND L6

=> 12 and 115

L16 992 L2 AND L15

=> 18 and 116

L17 22 L8 AND L16

=> d 117 11-22 ti

L17 ANSWER 11 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN trans-1,2-Dicyanocyclobutane

- L17 ANSWER 12 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Production of methacrylate and acrylate esters without polymerization
- L17 ANSWER 13 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Aluminum-plastic laminates
- L17 ANSWER 14 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Coatings for paper and cardboard
- L17 ANSWER 15 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Separation of methyl acetate and methanol from mixtures of. acetic acid and methyl chloride
- L17 ANSWER 16 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polymers of chlorine-substituted fluorobutadienes
- L17 ANSWER 17 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Ion-retardation for the separation of glycerol and sodium chloride in aqueous solution
- L17 ANSWER 18 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Reaction of acrylates and methylacrylates with organomagnesium compounds
- L17 ANSWER 19 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Nitration studies. XII. Nitrohalogenation of negatively substituted olefins with mixtures of dinitrogen tetroxide and halogens
- L17 ANSWER 20 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Allyl esters of some acids of trivalent arsenic and antimony, and an attempt at their copolymerization
- L17 ANSWER 21 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polymerization inhibition during acrylate purification
- L17 ANSWER 22 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Purifying materials such as methacrylic acid
- => d 117 12,21,22 ti
- L17 ANSWER 12 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Production of methacrylate and acrylate esters without polymerization
- L17 ANSWER 21 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polymerization inhibition during acrylate purification
- L17 ANSWER 22 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Purifying materials such as methacrylic acid
- => d 117 12,21,22 ti fbib abs
- L17 ANSWER 12 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Production of methacrylate and acrylate esters without polymerization
- AN 1967:473164 CAPLUS
- DN 67:73164
- TI Production of methacrylate and acrylate esters without polymerization

```
ΙN
    Hagemeyer, Hugh J., Jr.; Blood, Alden E.; Statman, Max
PA
    Eastman Kodak Co.
SO
    U.S., 4 pp.
    CODEN: USXXAM
DΤ
    Patent
    English
ĽΑ
FAN.CNT 1
                                  APPLICATION NO. DATE
    PATENT NO. KIND DATE
                          19670725 US
    US 3332974 ---- 19670725
                                                         19610718
PΙ
    The title process is carried out in distillation columns and reactors
    packed with a ceramic material which has been treated with an HNO3-H2SO4
    mixture, by treating methacrylamide (I) or acrylamide with an alc.
    in the presence of a strong aqueous acid solution Thus, aqueous I, prepared
    ml. acetone cyanohydrin and 3660 ml. of 100% H2SO4 was fed into the top of
    a column packed with unglazed porcelain Berl saddles. A total of 10,225
    ml. iso-BuOH was fed into a heater at the base of the column over 15.8
    hrs., and phenol and air were also fed into the reactor to inhibit
    polymerization A yield of 56.8 g. product per in.3 Berl saddles was
    obtained before the reactor column became plugged with polymer.
    The same process was carried out, using a column packed with Berl saddles
    which had been treated with a mixture of equal vols. of fuming sulfuric acid
     (20% SO3) and concentrated HNO3 for 12 hrs. A yield of 980 g. product per in.3
    Berl saddles was obtained from this column before the operation was
    terminated, and no sign of polymer was found on the
    packing. HCl and H3PO4 are also claimed as esterification
    catalysts, and benzyl alc., phenylethyl alc., cyclohexanol, cyclopentanol,
    and cyclobutanol as starting materials.
    ANSWER 21 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
L17
    Polymerization inhibition during acrylate purification
ΤI
    1957:5628 CAPLUS
AN
    51:5628
DN
OREF 51:1248b-d
    Polymerization inhibition during acrylate purification
IN
    Vaughan, Mervyn F.; Bellringer, Frederick J.
    Distillers Co. Ltd.
PA
DT
    Patent
LΑ
    Unavailable
FAN.CNT 1
                                        APPLICATION NO. DATE
                 KIND DATE
    PATENT NO.
     _____
                          19560410
                                        US
    US 2741583
PΙ
    Instead of purifying acrylates by use of expensive means for
    generation of NO2, it has been discovered that a metallic nitrite or a
    mixture of N oxides can be used to reduce polymerization in the
    still. The distillate produced, however, undergoes rapid
    polymerization, but this can be inhibited by governing the
     conditions closely. A solution (I) (50% by weight Me acrylate) with
    MeOH and EtCO2Me was vaporized and fed into a 66 + 2-in.
     fractionating column packed with stainless-steel gauze packing
    rings, and an aqueous solution of NaNO2 (II) (3 g./l.) at pH 6.0 and 71°
    fed into the system (1 atmospheric) through the top of the column. The
aqueous layer
    of the distillate was separated from the oily layer and returned to
     the column. The distillate was fed directly into a washing
     column and washed with a solution containing Na2SO3 and NaCl. The still ran
600
    hrs. without any evidence of polymer being formed. Replacement
     of the aqueous NaNO2 by a solution of any one of the alkaline earth-metal
```

or any of the other alkali-metal nitrites, or a mixture of N oxides produced

by the reaction of NaNO2 with dilute H2SO4, also inhibited **polymer** formation. I fed into a 40-plate stainless-steel, bubble-cap column at 2 gal./hr. and II fed into the top of the column at 12 gal./hr. and 44° and (pressure at the head of the column was 250 mm. and 332 mm. at the kettle) gave no serious **polymerization** after 420 hrs.

L17 ANSWER 22 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN

TI Purifying materials such as methacrylic acid

AN 1941:32661 CAPLUS

DN 35:32661

OREF 35:5133i,5134a-b

TI Purifying materials such as methacrylic acid

IN Barnes, Carl E.

PA Norton Co.

DT Patent

LA Unavailable

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
US 2241175 19410506 US

PI US 2241175 19410506 US

A method of purifying a monomeric polymerizable substance comprising an acid having the formula CH2:CRCOOH in which R is from the group of H, halogen and the methyl group involves the steps of distilling the substance and passing the vapor into a fractionating column and there causing the vapor and condensate to contact intimately with a packing presenting an extensive surface area of an inhibitor metal selected from the group consisting of Cu or Cr which is located wherever polymerization tends to occur materially, and causing the desired vapor to pass substantially free from the inhibitor and impurities to a cooling zone and there condensing the vapor and quickly cooling the condensate to a temperature at which polymerization will not take place.

# => d l17 1-10 ti

- L17 ANSWER 1 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Apparatus for handling of polymerizing substances
- L17 ANSWER 2 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Procedure for manufacture of (meth) acrylic acid esters
- L17 ANSWER 3 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for preparation of **packing** materials for liquid chromatography columns
- L17 ANSWER 4 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for cleaning of materials in packed towers
- L17 ANSWER 5 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for purification of isocyanatoalkyl (meth)acrylate substantially free from chlorine by distillation and dechlorination using epoxy compound and amine
- L17 ANSWER 6 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Resin composition and **distillation** methods for inhibiting **polymerization** of vinyl compound
- L17 ANSWER 7 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Preparation of hydrophilic column packings using organic polymers
- L17 ANSWER 8 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN

TISuppressing polymerization of acrylate and methacrylate monomers during rectification ANSWER 9 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN L17Acrylic acid purification and polymerization TIL17 ANSWER 10 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN TΙ Acrylic acid purification => d 117 1-10 ti fbib abs ANSWER 1 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN L17TIApparatus for handling of polymerizing substances AN2003:782784 CAPLUS 139:293789 DN Apparatus for handling of polymerizing substances TIKono, Michiyuki; Kaho, Yasuhiro; Taniguchi, Yoshiyuki; Marumoto, Takehiro IN Mitsubishi Rayon Co., Ltd., Japan PA SO Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF DT Patent LΑ Japanese FAN.CNT 1 APPLICATION NO. KIND DATE DATE PATENT NO. \_\_\_\_\_ \_\_\_\_\_ PIJP 2003284942 A2 20031007 JP 2002-92510 20020328 JP 2002-92510 20020328 The title apparatus is composed of packing materials coated with, AB e.g., fluoropolymers. It is used for prevention of growth of polymerizing substances, e.g., (meth)acrylic acid in distillation ANSWER 2 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN L17 Procedure for manufacture of (meth) acrylic acid esters TΙ 2002:462390 CAPLUS ΑN 137:33676 DN Procedure for manufacture of (meth)acrylic acid esters ТT Martin, Friedrich-Georg; Nestler, Gerhard; Schroeder, Juergen IN BASF AG, Germany PA Ger. Offen., 4 pp. SO CODEN: GWXXBX DTPatent LΑ German FAN.CNT 1 APPLICATION NO. DATE PATENT NO. KIND DATE -----\_\_\_\_ 20020620 DE 2000-10063176 20001218  $_{
m PI}$ DE 10063176 A1 WO 2001-EP14903 20011217 WO 2002050015 A120020627 W: US RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR DE 2000-10063176A 20001218 20030924 EP 2001-984858 20011217 EP 1345887 A1 . R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, FI, CY, TR DE 2000-10063176A 20001218 WO 2001-EP14903W 20011217 US 2004030179 A1 20040212 US 2003-433614 20030617 DE 2000-10063176A 20001218 WO 2001-EP14903W 20011217 (meth)acrylic acid esters (of mol. weight >200) are obtained by AΒ

esterification of (meth)acrylic acid with alcs. in the presence

of  $\geq 1$  acid catalyst,  $\geq 1$  polymerization inhibitor, and an organic solvent, which forms an azeotrope with water, whereby the the mixture is heated to the b.p. in an apparatus with a distillation unit, column and condenser, the azeotrope is distilled off and the organic solvent is recirculated to the column at least partially contacting a copper-containing material, such as distillation column packings or separation efficient fittings. Thus, acrylic acid 2380, tripropylene glycol 2880, cyclohexane 2300, p-toluenesulfonic acid 120, 50% phosphinic acid 9.4, and hydroquinone monomethylether 4.7 parts were mixed in a 10-L-reactor with a double layer heating and distillation column (5 + 60 cm). The reaction water formed was **distilled** off as azeotrope with cyclohexane, whereby after after condensation the organic phase formed was separated and recirculated to the column, which was filled with copper Raschig-rings at the top and glass rings under it. Within 8 h, 546 parts of water was separated, so that a 97% yield of esterification took place and no polymerization was observed in the column.

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L17 ANSWER 3 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
```

- TI Method for preparation of **packing** materials for liquid chromatography columns
- AN 2000:669496 CAPLUS
- DN 133:208334
- TI Method for preparation of **packing** materials for liquid chromatography columns
- IN Song, Soo-suk; Kim, Ho-hyun; Yeu, Kyu-dong
- PA Samyang Co., S. Korea
- SO Repub. Korea, No pp. given CODEN: KRXXFC
- DT Patent
- LA Korean
- FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	KR 9709234	В1	19970609	KR 1993-27285	19931210
				KR 1993-27285	19931210

- The column packing material for liquid chromatog., which has a improved mech. strength and resolution performance, is prepared by suspension polymerization via addition of emulsifiers, a salting out agent, and initiator to a 30-60% pentaerythritol mono-, di-, tri(meta) acrylate mixture, and 1-20% cyclohexane di-Me di(meta) acrylate derivs., and 39-50% glycidyl (meta)acrylate in distilled water and organic solvent.
- L17 ANSWER 4 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for cleaning of materials in packed towers
- AN 1999:714922 CAPLUS
- DN 131:324471
- TI Method for cleaning of materials in packed towers
- IN Suzuki, Masao; Nomura, Kiyohito; Ikemori, Shinji
- PA Yuken Kogyo Co., Ltd., Japan; Idemitsu Kosan Co., Ltd.
- SO Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese
- FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 11309363	A2	19991109.	JP 1998-363399	19981221
				JP 1997-353819	19971222

AB Method for cleaning of inorg. materials packed in a tower is claimed. Gels formed as byproducts in preparation of polymerizable compds. by gas-liquid contact reaction in the tower are adhered on the inorg. packings. Used packings are (1) immersed in monoterpene

solvents for the gels to swell, (2) then barrel treated with a monoterpene-containing emulsion for removal of the separated gel, and (3) the **packings** are rinsed and heat-dried. The **packings** can be cleaned to a recycling level.

- L17 ANSWER 5 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Method for purification of isocyanatoalkyl (meth)acrylate substantially free from chlorine by distillation and dechlorination using epoxy compound and amine
- AN 1999:530982 CAPLUS
- DN 131:158089
- TI Method for purification of isocyanatoalkyl (meth)acrylate substantially free from chlorine by distillation and dechlorination using epoxy compound and amine
- IN Misu, Naoaki; Matsuhira, Shinya; Kihara, Muneyo; Ohnishi, Yutaka
- PA Showa Denko K. K., Japan
- SO Jpn. Kokai Tokkyo Koho, 7 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese
- FAN CNT 1

r AN.	CM.L	T																
	PA?	CENT N	10.		KI	ND	DATE			AF	PLIC	CATI	ON N	ο.	DATE			
PΙ	JР	11228	3523		Α	2	1999	0824		JF	199	98-2	5493		1998	0206		
	CA	22613	324		Α	A	1999	0806		C <i>P</i>	199	99-2	2613	24	1999	0205		
										JE	19:	98-2	5493	Α	1998	0206		
	ΕP	93621	1.4		Α	2	1999	0818		EF	19	99-1	0231	8	1999	0205		
	EP	93621			Α	-	1999											
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	, NL,	SE,	MC,	PT,
			IE,	SI,	LT,	LV,	FI,	RO										
										JE	19	98-2	5493	Α	1998	0206		
	US	62459	935		В	1	2001	0612		US	19	99-2	4570	7	1999	0208		
										JE	19	98-2	5493	Α	1998	0206		
										US	19	98-1	0152	7PP	1998	0923		

- AB Isocyanatoalkyl (meta)acrylates substantially free from hydrolytic chlorine are prepared by purification which involves treatment of (A)
  - isocyanatoalkyl acrylate containing isocyanatoalkyl 2-chloropropionate or (B) isocyanatoalkyl methacrylate containing isocyanatoalkyl 2-methyl-2-chloropropionate with an epoxy-containing compound and amine/or imidazole until isocyanatoalkyl 2-chloropropionate or 2-methyl-2-chloropropionate is no longer present. The purified isocyanatoalkyl (meta)acrylate is useful as a raw material for photoresists (active ray-curable resins) suitable for electronic or elec. parts which is not compatible with chlorine. Thus, 2-isocyanatoethyl methacrylate (I) containing 381 ppm hydrolytic chlorine 300, epoxidized fatty plasticizer (mol. weight .apprx.100 and iodine value 7) containing 6.1% oxirane oxygen 1.7, 2,6-di-tert-butyl-4-methylphenol 0.3, and triethylenetetramine 0.11 g were stirred in a glass reaction vessel at 60° and .apprx.1.3 kPa and distilled at 85° to give 220 g I containing 29 ppm hydrolytic chlorine. Phenothiazine (0.15 g) was added the purified I (150 g) and the resulting mixture was distilled at  $70^{\circ}$ (column bottom temperature 81°) and .apprx.0.7 kPa with a series of two glass columns packed with Dixon packings to give 53 g I in which no hydrolytic chlorine was detected.
- L17 ANSWER 6 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Resin composition and distillation methods for inhibiting polymerization of vinyl compound
- AN 1999:260038 CAPLUS
- DN 130:312219
- TI Resin composition and distillation methods for inhibiting polymerization of vinyl compound

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ΙN
     Tanaka, Kazumi
     Mitsubishi Gas Chemical Company, Inc., Japan
PA
SO
     Eur. Pat. Appl., 15 pp.
     CODEN: EPXXDW
DT
     Patent
     English
LA
FAN.CNT 1
                                           APPLICATION NO.
     PATENT NO.
                      KIND DATE
                                                            DATE
                            19990421
                                           EP 1998-119189
PΙ
     EP 909767
                      A1
                                                            19981012
                            20040204
     EP 909767
                      В1
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
                                           JP 1997-283637 A 19971016
                                           JP 1998-70449 A 19980319
                       A2
                                           JP 1997-283637
                                                            19971016
     JP 11116819
                            19990427
                       A2
                            19991005
                                           JP 1998-70449
                                                            19980319
     JP 11269128
                            20010529
                                           US 1998-172197
     US 6239199
                      _{\rm B1}
                                                            19981014
                                           JP 1997-283637 A 19971016
                                           JP 1998-70449 A 19980319
     There are disclosed (1) a resin composition for inhibiting polymerization of
AB
     a vinyl compound, which composition comprises a resin and 0.1 to 30% by weight
based
     on the resin, of a polymerization inhibitor for the vinyl compound; (2) a
     resin molding for inhibiting polymerization of a vinyl compound
     comprising the above resin composition; (3) packing for inhibiting
     polymerization of a vinyl compound in distillation tower, which
     packing comprises the above resin composition or resin molding; (4) a
     method for inhibiting polymerization of a vinyl compound comprising
     contacting the vinyl compound with the resin composition; and (5) a method for
     inhibiting polymerization of a vinyl compound comprising contacting the
     vinyl compound in a distillation tower with the packing
     therein comprising the above resin composition By using the resin composition
for
     inhibiting polymerization of the vinyl compound of the present invention
     as a construction material of an inner wall of a reactor, distillation
     equipment and piping and the packing in the distillation
     tower, the vinyl compound can be produced stably by adding a small amount of
     the polymerization inhibitor for the vinyl compound without adding a
     large amount thereof, differently from the prior methods. Thus, a composition
as
     polymerization inhibitor of 2-hydroxyethyl methacrylate in a
     distillation tower was made from FY 6C containing 5% phenothiazine (Antage
     TDP).
RE.CNT 10
              THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD
              ALL CITATIONS AVAILABLE IN THE RE FORMAT
     ANSWER 7 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
L17
ΤI
     Preparation of hydrophilic column packings using organic
     polymers
ΑN
     1990:461839 CAPLUS
     113:61839
DN
ΤI
     Preparation of hydrophilic column packings using organic
     polymers
     Sulc, Jiri; Linek, Vaclav; Kralicek, Jaroslav; Krivsky, Zdenek; Sinkule,
IN
     Jiri; Raschig, Gert
PA
     Czech.
     Czech., 4 pp.
SO
     CODEN: CZXXA9
DΤ
     Patent
LΑ
     Czech
FAN.CNT 1
                      KIND DATE
                                           APPLICATION NO. DATE
     PATENT NO.
```

PI CS 263649 B1 19890414 CS 1986-8417 19861120 CS 1986-8417 19861120

The packing materials comprise molded polymer melts in which 1-15 weight% hydrophilic polymer having a contact wetting angle <40° is added to a hydrophobic polymer (e.g., polypropylene) having a contact wetting angle >60°. The hydrophilic polymer has the same or lower m.p. and lower viscosity than the hydrophobic polymer. Ethylene-vinyl acetate copolymer containing ≥40% vinyl acetate, and/or low-mol. weight sulfonated polystyrene, sulfonated copolymers of styrene with ethylene or other vinyl monomers, ethylenesulfonic acid copolymers, partially hydrolyzed polymers and copolymers of acrylonitrile (e.g., acrylonitrile-acrylamide copolymer), and copolymers of alkylacrylates, preferably Me acrylates, with acrylic or methacrylic acid are used as the hydrophilic polymers. The hydrophilic packings can be used in columns for sorption, distillation, chemisorption, cooling, etc., which operate with aqueous phases.

- L17 ANSWER 8 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Suppressing **polymerization** of **acrylate** and methacrylate monomers during rectification.
- AN 1975:459753 CAPLUS
- DN 83:59753
- TI Suppressing polymerization of acrylate and methacrylate monomers during rectification
- IN Kouril, Vladimir
- SO Czech., 2 pp.

CODEN: CZXXA9

DT Patent

LA Czech

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	CS 153894	В	19740329	CS 1972-1225	19720225
				CS 1972-1225	19720225

AB Ceramic column packings were treated with a 3:1 mixture of H2SO4 and HNO3, washed, dried, pressure-impregnated with a 15% CuCl2 solution and dried at 120°. The equipment remained clean after a 2-hr exptl. distillation of a 7:3 mixture of Bu and Me acrylate, while untreated columns became clogged with polymers despite the presence of a polymerization inhibitor.

- L17 ANSWER 9 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Acrylic acid purification and polymerization
- AN 1970:467209 CAPLUS
- DN 73:67209
- TI Acrylic acid purification and polymerization
- IN Bashaw, Robert N.
- PA Dow Chemical Co.
- SO Brit., 3 pp. CODEN: BRXXAA
- DT Patent
- LA English
- FAN.CNT 1

T 1 114 .	OIL I			
	PATENT NO. KIND	DATE	APPLICATION NO.	DATE
ΡI	GB 1197558	19700708	GB	19690328
	FR 2040874		FR	

AB Acrylic acid (I) suitable for the preparation of high mol. weight, flocculant grade polymers was prepared by distilling inhibited I through a Cu packing, recovering the

distillate in water at <20°, and treating the aqueous solns. of I distillate with a cation exchange resin, e.g. sulfonated styrene-divinylbenzene copolymers, to reduce the Cu content of the

```
distillate solution to <1 ppm.
    ANSWER 10 OF 22 CAPLUS COPYRIGHT 2004 ACS on STN
L17
     Acrylic acid purification
ΤI
     1970:78435 CAPLUS
AN
DN
     72:78435
TI
     Acrylic acid purification
ΙN
     Bashaw, Robert N.
PA
     Dow Chemical Co.
     U.S., 2 pp.
SO
     CODEN: USXXAM
DT
     Patent
LΑ
     English
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                          APPLICATION NO. DATE
                      ____
                           _____
                                           _____
                                                           _____
PΙ
     US 3493471
                       Α
                            19700203
                                           US 1967-687153
                                                            19671201
                            19691014
                                           BE 1969-731457
                                                            19690414
     BE 731457
                      Α
                                           US 1967-687153 19671201
AΒ
     Quinone- or thiazine-inhibited acrylic acid (I) was purified to
     polymerization by distillation through a Cu packing.
     overhead vapors were quenched with deionized water at .apprx.10°.
     The resulting I solution was contacted with an ion exchange resin or
     achelating resin to remove the Cu cations. The resulting I could then be
     polymerized in the presence of FeSO4 and \alpha,\alpha'-
     azobisisobutyronitrile, yielding a high-mol.-weight, flocculant-grade poly(
     acrylic acid).
=> d his
     (FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)
     FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004
           3818 GAS INJECTION
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=> 118 and 115

L19

810 L18 AND L15

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L1
L2
         402843 ACRYL?
L3
             20 L1 AND L2
L4
        1868920 POLYM?
L5
            159 L1 AND L4
L6
          72711 PACKING
L7
             18 L1 AND L6
\Gamma8
         353964 DISTILL?
L9
             38 L1 AND L8
L10
              1 L4 AND L9
          15009 CORRUG?
L11
L12
              2 L1 AND L11
L13
            542 L6 AND L11
             44 L4 AND L13
L14
L15
          10957 L4 AND L6
L16
            992 L2 AND L15
L17
             22 L8 AND L16
=> styrene
        251507 STYRENE
          4248 STYRENES
L18
        252537 STYRENE
                  (STYRENE OR STYRENES)
```

=> 18 and 119

L20 24 L8 AND L19

- => d 120 11-24 ti
- L20 ANSWER 11 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Packed towers reduce cost
- L20 ANSWER 12 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Acrylic acid purification and polymerization
- L20 ANSWER 13 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Multiple detectors for molecular weight and composition analysis of copolymers by gel permeation chromatography
- L20 ANSWER 14 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Aluminum-plastic laminates
- L20 ANSWER 15 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Separation of methyl acetate and methanol from mixtures of. acetic acid and methyl chloride
- L20 ANSWER 16 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Coal-tar styrene
- L20 ANSWER 17 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polymers of chlorine-substituted fluorobutadienes
- L20 ANSWER 18 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Reactions catalyzed by acids or bases
- L20 ANSWER 19 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Ion-retardation for the separation of glycerol and sodium chloride in aqueous solution
- L20 ANSWER 20 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI 2-Phenylbutadiene from  $\alpha$ -methylstyrene
- L20 ANSWER 21 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI X-ray characterization of new isotactic polymers
- L20 ANSWER 22 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Allyl esters of some acids of trivalent arsenic and antimony, and an attempt at their copolymerization
- L20 ANSWER 23 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Derivatives of dicyclopentadienyliron
- L20 ANSWER 24 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Distillation of polymerizable vinyl aromatic compounds such as styrene and ethylbenzene
- => d 120 12, 24 ti fbib abs
- L20 ANSWER 12 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Acrylic acid purification and polymerization
- AN 1970:467209 CAPLUS
- DN 73:67209
- TI Acrylic acid purification and polymerization
- IN Bashaw, Robert N.
- PA Dow Chemical Co.

SO Brit., 3 pp. CODEN: BRXXAA

DT Patent

LA English

FAN.CNT 1

AB Acrylic acid (I) suitable for the preparation of high mol. weight, flocculant grade polymers was prepared by distilling inhibited I through a Cu packing, recovering the distillate in water at <20°, and treating the aqueous solns. of I distillate with a cation exchange resin, e.g. sulfonated styrene—divinylbenzene copolymers, to reduce the Cu content of the distillate solution to <1 ppm.

L20 ANSWER 24 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN

TI Distillation of polymerizable vinyl aromatic compounds such as styrene and ethylbenzene

AN 1941:32637 CAPLUS

DN 35:32637

OREF 35:5131d-e

TI Distillation of polymerizable vinyl aromatic compounds such as styrene and ethylbenzene

IN Dreisbach, Robert R.; Pierce, James E.

PA The Dow Chemical Co.

DT Patent

LA Unavailable

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI US 2240764

19410506 U

AB Apparatus is described, and a mode of operation which involves distillation through a distilling column containing a packing comprising a substantially insol. agent effective in inhibiting polymerization of the vinyl aromatic compound, the packing extending upward in the column so that the distilling vapors must pass through it.

# => d 120 1-10 ti

- L20 ANSWER 1 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis and Hydrolysis of Methyl Acetate by Reactive **Distillation** Using Structured Catalytic **Packings:** Experiments and Simulation
- L20 ANSWER 2 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Characterization and catalytic activity of a novel resin-type catalyst
- L20 ANSWER 3 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis of MTBE. Influence of crosslinking and **polymer** content on activity and selectivity of **polymer** carrier ion exchange catalysis
- L20 ANSWER 4 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Influence of crosslinking and **polymer** content on activity and selectivity of **polymer**/carrier ion exchange catalysts for the synthesis of MTBE
- L20 ANSWER 5 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Preparation of catalytic polymer/ceramic ion-exchange packings for reactive distillation columns

- L20 ANSWER 6 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Study of fouling and antifouling method for gasoline fractionator in ethylene plant
- L20 ANSWER 7 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Preparation of hydrophilic:column packings using organic polymers
- L20 ANSWER 8 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Application of a new ordered packing F1 in styrene rectification columns
- L20 ANSWER 9 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Styrene from pyrolysis gasoline
- L20 ANSWER 10 OF 24 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Catalyst structure

=> logoff hold

COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	133.88	134.09
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-14.55	-14.55

SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 08:09:39 ON 26 FEB 2004

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Welcome to STN International! Enter x:x

LOGINID:SSSPTA1623PAZ

# PASSWORD:

\* \* \* \* \* RECONNECTED TO STN INTERNATIONAL \* \* \* \* \* \* SESSION RESUMED IN FILE 'CAPLUS' AT 08:37:52 ON 26 FEB 2004 FILE 'CAPLUS' ENTERED AT 08:37:52 ON 26 FEB 2004 COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	133.88	134.09
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-14.55	-14.55

=> d his

(FILE 'HOME' ENTERED AT 06:10:31 ON 26 FEB 2004)

FILE 'CAPLUS' ENTERED AT 06:10:44 ON 26 FEB 2004

- L1 3818 GAS INJECTION
- L2 402843 ACRYL?
- L3 20 L1 AND L2

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1868920 POLYM?
L4
            159 L1 AND L4
L5 -
          72711 PACKING
L6
             18 L1 AND L6
L7
         353964 DISTILL?
\Gamma8
L9
             38 L1 AND L8
L10
              1 L4 AND L9
          15009 CORRUG?
L11
              2 L1 AND L11
L12
L13
            542 L6 AND L11
L14
             44 L4 AND L13
          10957 L4 AND L6
L15
L16
            992 L2 AND L15
L17
            22 L8 AND L16
         252537 STYRENE
L18
            810 L18 AND L15
L19
            24 L8 AND L19
L20
=> packing layer
         69768 PACKING
          6972 PACKINGS
         72711 PACKING
                 (PACKING OR PACKINGS)
       1072420 LAYER
        479972 LAYERS
       1315819 LAYER
                 (LAYER OR LAYERS)
L21
           210 PACKING LAYER
                 (PACKING(W) LAYER)
=> 121 and 111
L22
             9 L21 AND L11
=> d 122 1-9 ti
L22 ANSWER 1 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
     Preparation of wire-mesh honeycomb coated with alumina-encapsulated
     aluminum layer
L22 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
    Modifications to structured packings to increase their capacity
TΤ
    ANSWER 3 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
L22
TI
     Regular packing for heat- and mass-transfer apparatus
    ANSWER 4 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
L22
     Influence of structural parameters of corrugated plate packing
TΙ
     on its performances
L22
    ANSWER 5 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
ΤI
     Configuration analysis and improvement measures of structured packing in
     desorption column
    ANSWER 6 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
L22
     Chemical process tower deentrainment assembly for vapor-liquid mass
TΙ
     transfer
L22 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
     Gas-liquid contacting packed columns with specified packing material
     blocks and liquid distributors
L22 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
```

Modeling and measurement of gas flow distribution in corrugated

sheet structure packings

- L22 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Ordered column packing for diffusion processes

### => d 122 1-9 ti fbib abs

- L22 ANSWER 1 OF 9. CAPLUS COPYRIGHT 2004 ACS on STN
- TI Preparation of wire-mesh honeycomb coated with alumina-encapsulated aluminum layer
- AN 2003:440721 CAPLUS
- DN 139:183598
- TI Preparation of wire-mesh honeycomb coated with alumina-encapsulated aluminum layer
- AU Yang, K. S.; Jiang, Z. D.; Chung, J. S.
- CS Department of Chemical Engineering, Pohang University of Science and Technology, Pohang, 790-784, S. Korea
- SO Advances in Science and Technology (Faenza, Italy) (2003), 33(10th International Ceramics Congress, 2002, Part D), 403-410 CODEN: ASETE5
- PB Techna
- DT Journal
- LA English
- AB We developed a new design of wire-mesh honeycomb (WMH) which is a metallic monolith constructed by packing layers of flat and corrugated wire-mesh sheets alternatively within a frame. Aluminum powder was coated on the wire meshes using electrophoretic deposition (EPD). The Al particles were well adhered onto the substrate and the thickness of coated layer was .apprx.100μm. Thin Al2O3 layer was formed in the outer surface of the Al particles after calcination, completely encapsulating each Al particles. It has porous structure with a large surface area. The Al/Al2O3-coated WMH was wash-coated with Pt/TiO2 catalyst and applied for catalytic combustion of benzene. It showed better activity compared with the conventional ceramic honeycomb.
- RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT
- L22 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Modifications to structured packings to increase their capacity
- AN 2003:156505 CAPLUS
- DN 138:306028
- TI Modifications to structured packings to increase their capacity
- AU Bender, P.; Moll, A.
- CS Linde AG, Hollriegelskreuth / Munich, Germany
- SO Chemical Engineering Research and Design (2003), 81(A1), 58-67 CODEN: CERDEE; ISSN: 0263-8762
- PB Institution of Chemical Engineers
- DT Journal
- LA English
- AB The hydraulic performance of structured packed columns is characterized by parameters such as loading point, flooding point, and dry and wet pressure-drop. These parameters can be pos. influenced by optimizing the transition areas between two adjacent packing layers.

Tests have been done in a two-phase test rig with a rectangular column made of perspex. The test medium is a liquid hydrocarbon in counterflow to saturated nitrogen gas. The performance of different packing structures at the layer-to-layer transition zone and the addition of vane-type elements between two packing layers was evaluated. Several

packing modifications were tested and the results were compared with those of a non-modified packing. The intentions of this work are: to rank the modifications of structured packings with respect to the achieved capacity increase; to determine whether flooding is initiated in the core of the

corrugated packing or in the transition area in spite of the modifications made on the lower or top side of the packing sheets; and to find out how the loading point is influenced by the modifications.

RE.CNT 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 3 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

TI Regular packing for heat- and mass-transfer apparatus

AN 2003:82738 CAPLUS

DN 138:305876

TI Regular packing for heat- and mass-transfer apparatus

IN Zibert, G. K.; Kashchitskii, Yu. A.; Kulikova, S. N.

PA Dochernee Otkrytoe Aktsionernoe Obshchestvo "Tsentral'noe Konstruktorskoe Byuro Nefteapparatury" Otkrytogo Aktsionernogo Obshchestva "Gazprom", Russia

SO Russ., No pp. given

CODEN: RUXXE7

DT Patent

LA Russian

FAN.CNT 1

PΙ

PATENT NO. KIND DATE APPLICATION NO. DATE -----\_\_\_\_\_ \_\_\_\_ \_\_\_\_\_ \_\_\_\_\_ 20020910 20010115 RU 2188706 C1 RU 2001-101106 RU 2001-101106 20010115

The packing consists of vertical corrugated sheets with AB projecting corrugations in contact; each corrugation is made with inversely concave elements which are located at points of intersection of parallel lines with edges of corrugations. Transversal notches are made on corrugations along parallel lines at points of intersection with edges of corrugations; inversely concave elements in form of trapezia which are so bent that a bent line lies on an inversely concave edge of corrugation. Corrugations along parallel lines at points of their intersection with fins of corrugations are provided with transversal notches; inversely concave elements in form of triangles whose bases are equal to a length of notches and vertices located on edges of corrugations are made at each notch; they are so bent that lines of bend lie on inversely concave edges of corrugations; notches with inversely concave elements are also made on corrugations on back side of sheet. Parallel lines intersecting the edges of corrugations at point of location of inversely concave elements are located at an angle of 60-90° relative to the edge of corrugations. Perforated sheets or meshes are mounted between corrugated sheets; sizes of their meshes ensure a continuous liquid flow. The arrangement enhances efficiency of mass and heat exchange due to turbulization of flows inside the packing layer. The packing is suitable for processes of rectification, absorption, purification, and drying of natural gas, as well as mixing of liquid and gas flows, separating phases in separating units,

as a contact element in mixing condensers for all technol. processes of petroleum and gas industry.

- L22 ANSWER 4 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Influence of structural parameters of **corrugated** plate packing on its performances
- AN 2002:357711 CAPLUS
- DN 137:21820
- TI Influence of structural parameters of **corrugated** plate packing on its performances
- AU Jia, Shaoyi; Sun, Yongli; Zhao, Jinduo; Wu, Songhai; Xing, Yikun
- CS School of Chemical Engineering, Tianjin University, Tianjin, 300072, Peop. Rep. China
- SO Huagong Xuebao (Chinese Edition) (2002), 53(4), 364-368

CODEN: HUKHAI; ISSN: 0438-1157

- PB Huaxue Gongye Chubanshe, Huagong Xuebao Bianjibu
- DT Journal
- LA Chinese
- AB The effect of structural parameters of corrugated plate packing on its performance of axial mixing in liquid phase and hydrodynamic and mass transfer performance is investigated in a 300 mm diameter packing tower. The axial backmixing parameter, Pe, increases with increasing opening ratio,  $\phi$ , and the inclination angle,  $\beta$ , and with decreasing the unit height of the packing. The axial backmixing parameter, Pe, increases with increasing opening ratio, , and the inclination angle, , and with decreasing the unit height of the packing, H, and that the pressure drop of packing layer,  $\Delta p/Z$ , decreased with increasing  $\phi$  and H and with decreasing  $\beta$  and that the height of mass transfer unit HOG decreased with increasing  $\beta$  and with decreasing H, and that effect  $\phi$  on the HOG had a suitable value. Through regressing exptl. data, the correlation of the Pe and  $\Delta p/Z$ , and the HOG were obtained. The results can be used in the development and design of corrugated plate packing.
- L22 ANSWER 5 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Configuration analysis and improvement measures of structured packing in desorption column
- AN 1999:356178 CAPLUS
- DN 131:60408
- TI Configuration analysis and improvement measures of structured packing in desorption column
- AU Chen, Yuping
- CS Hubei Zaoyang Chemical Industry General Corp., 441200, Peop. Rep. China
- SO Huafei Gongye (1999), 26(2), 56-57 CODEN: HUGOFO; ISSN: 1006-7779
- PB Huafei Gongye Bianjibu
- DT Journal
- LA Chinese
- AB Problems existed in configuration design of structured packing (250Y metal plate corrugated structured packing) in desorption column were discussed, such as improper design and setting of fluid distributor, improper height distribution of packing layers, etc.

  Proposals were put forward to improve fluid distributor and height of packing and column.
- L22 ANSWER 6 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Chemical process tower deentrainment assembly for vapor-liquid mass transfer
- AN 1998:87660 CAPLUS
- DN 128:129577
- TI Chemical process tower deentrainment assembly for vapor-liquid mass transfer
- IN Lee, Adam T.; Wu, Kuang; Burton, Larry; Fan, Leon
- PA Koch Enterprises, Inc., USA
- SO PCT Int. Appl., 25 pp. CODEN: PIXXD2
- DT Patent
- LA English
- FAN CNT 1

FAN.	CNT	1																
	PATENT NO.			KI	ND	DATE			Α	PPLI	CATI	N NC	٥.	DATE				
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PI	I WO 9803258		A	1	19980129			WO 1997-US12767		67	19970722							
		w:	AL,	AM,	AT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CU,	CZ,	DE,
			DK,	EE,	ES,	FI,	GB,	GE,	GH,	HU,	IL,	IS,	JP,	KE,	KG,	KP,	KR,	ΚZ,
			LC,	LK,	LR,	LS,	LT,	LU,	LV,	MD,	MG,	MK,	MN,	MW,	MX,	NO,	NZ,	PL,
			PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TR,	TT,	UA,	UG,	UZ,
			VN,	YU,	ZW,	AM,	ΑZ,	BY,	KG,	ΚZ,	MD,	RU,	ТJ,	TM				

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RW: GH, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, DE, DK, ES, FI, FR,
        GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA,
        GN, ML, MR, NE, SN, TD, TG
                                       US 1996-685482 A 19960724
US 5762668
                                       US 1996-685482
                  Α
                       19980609
                                                        19960724
                                       CA 1997-2261146 19970722
CA 2261146
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                                       US 1996-685482 A 19960724
AU 9739613
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                       19980210
                                       AU 1997-39613
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AU 713193
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                       19991125
                                       US 1996-685482 A 19960724
                                       WO 1997-US12767W 19970722
ZA 9706490
                  Α
                       19980219
                                       ZA 1997-6490
                                                        19970722
                                       US 1996-685482 A 19960724
EP 914204
                       19990512
                                       EP 1997-936986
                                                        19970722
                  Α1
   R: BE, CH, DE, ES, FR, GB, IT, LI, NL
                                       US 1996-685482 P 19960724
                                       WO 1997-US12767W 19970722
BR 9710512
                       20000111
                                       BR 1997-10512
                                                        19970722
                                       US 1996-685482 A 19960724
                                       WO 1997-US12767W 19970722
                                       JP 1998-507169
JP 2001506913
                  T2
                       20010529
                                                        19970722
                                       US 1996-685482 A 19960724
                                       WO 1997-US12767W 19970722
TW 384235
                       20000311
                  В
                                       TW 1997-86110533 19970724
                                       US 1996-685482 A 19960724
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AB A deentrainment assembly for a mass transfer tray is described for a chemical process tower. The apparatus comprises a structured packing layer (e.g., corrugated sheet packing) assembled with a second type of packing layer (e.g., wire mesh) provided adjacent the underside of a mass transfer tower tray. A metal or plastic grid secures the assembly to the underside of the tray. The dual layer reduces liquid entrainment in ascending vapor flow and provides an addnl. region for mass transfer. In an example, the trays were used in distillation columns.

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

TI Gas-liquid contacting packed columns with specified packing material blocks and liquid distributors

AN 1995:682713 CAPLUS

DN 123:59874

TI Gas-liquid contacting packed columns with specified packing material blocks and liquid distributors

IN Yoshimatsu, Yukyoshi; Harada, Susumu; Someya, Kazuo

PA Hitachi Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI ·	JP 07080279	<b>A</b> 2	19950328	JP 1993-229851	19930916
				JP 1993-229851	19930916

AB In the title packed column comprising multistage packing layers formed from corrugated thin-layered articles having alternately arranged wave-shaped grooves between adjacent articles, and contacting a descending liquid with an ascending gas, the packing materials are divided into plural blocks, the outer walls of the packing materials are adjoined with the inner wall of the column, and approx.

L-shaped liquid distributors having their vertical sides arranged between the outer circumferences of the packing materials and the inner wall of

the column, and their horizontal sides arranged between the packing material blocks resp. Deformation of packing materials is prevented; separation efficiency is increased; pressure loss is lowered. The packed columns are especially suitable for rectification and cooling columns in air separation apparatus for separating O, N and Ar from air.

- L22 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Modeling and measurement of gas flow distribution in corrugated sheet structure packings
- AN 1994:167768 CAPLUS
- DN 120:167768
- TI Modeling and measurement of gas flow distribution in corrugated sheet structure packings
- AU Stoter, F.; Olujic, Z.; de Graauw, J.
- CS Lab. Process Equip., Delft Univ. Technol., Delft, 2628 CA, Neth.
- SO Chemical Engineering Journal (Amsterdam, Netherlands) (1993), 53(1), 55-66 CODEN: CMEJAJ; ISSN: 0300-9467
- DT Journal
- LA English
- AB A math. model and calcn. procedure are developed for the gas flow distribution in channels formed between tightly packed, corrugated , unperforated metal sheets. The model is a discrete cell model based on average mass, momentum and energy balance equations for each of numerous crossings of gas flow channels, with characteristic friction factors for gas inlet, bulk zone and wall zone as model parameters, which can be easily obtained from pressure drop measurements for each type and size of structured packing. The model enables prediction of velocity profiles leaving an element of packing layer consisting of segments of unperforated, structure packing. It is also suited for perforated packings which under operating (wetted) conditions function as a closed surface packing.
- L22 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Ordered column packing for diffusion processes
- AN 1990:481069 CAPLUS
- DN 113:81069
- TI Ordered column packing for diffusion processes
- IN Braun, Vlastimil
- PA Czech.
- SO Czech., 5 pp.
  - CODEN: CZXXA9
- DT Patent
- LA Czech
- FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PΙ	CS 263904	В1	19890512	CS 1987-3265	19870507
				CS 1987-3265	19870507

AB The title packing consists of parallel offset packing layers which fill the complete cross section of the column, each layer being made substantially from vertical parallel walls. The vertical walls are made from material having capillary properties, e.g., glass fibers, and are equipped with supporting tape at the top and bottom edges and, optionally, also in the middle. Corrugated vertical spacers made from solid material (e.g., metals, plastics, or ceramics) are fastened to the supporting tape at uniform spacings. The spacers are equipped with openings at least in the top part, and have bent spaced-apart holes in the longitudinal edges. Optionally, the spacers between the top and bottom supports are connected with ≥1 addnl. spacers.

COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 169.79 170.00

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

CA SUBSCRIBER PRICE

SINCE FILE TOTAL ENTRY SESSION 20.79 -20.79

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STN INTERNATIONAL SESSION SUSPENDED AT 08:46:35 ON 26 FEB 2004

Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID: SSSPTA1623PAZ

## PASSWORD:

\* \* \* \* \* RECONNECTED TO STN INTERNATIONAL \* \* \* \* \* \* SESSION RESUMED IN FILE 'CAPLUS' AT 09:30:59 ON 26 FEB 2004 FILE 'CAPLUS' ENTERED AT 09:30:59 ON 26 FEB 2004 COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

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ENTRY	SESSION
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SINCE FILE	TOTAL
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(RY OR RIES)

=> 18 and 123

L24 31 L8 AND L23

=> d 124 21-31 ti

- L24 ANSWER 21 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Some reactions of an optically active diazonium salt and of an optically active diacyl peroxide
- L24 ANSWER 22 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Substituted ethylenediamines
- L24 ANSWER 23 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Tetraalkyllead compounds
- L24 ANSWER 24 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Mixed mercaptal-acetals
- L24 ANSWER 25 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Lubricating-oil detergent
- L24 ANSWER 26 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Report of Committee XVII-wood preservation
- L24 ANSWER 27 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Report of Committee IV on preservatives
- L24 ANSWER 28 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Report of Committee 4 on [wood] preservatives
- L24 ANSWER 29 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Report Chief Inspector of Bureau for Safe Transportation of Explosives and Other Dangerous Articles
- L24 ANSWER 30 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI The Strength of Treated Timber. Report of the Committee on Wood Preservation of the Am. Railway English and Maintenance of Way Assn
- L24 ANSWER 31 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Modern Timber Preservation

=> rmax

L25 1447 RMAX

=> d 124 10-20 ti

- L24 ANSWER 10 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Reactions with microorganisms. XII. Stereospecificity of the reduction of the double bond in (  $\pm$  )-9-methyl- $\Delta 4$ -octalin-3,8-dione with Curvularia falcata
- L24 ANSWER 11 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthetic reactions of dimethylformamide. XVII. Preparation of acylmalonic dialdehydes by formylation of trimethinium salts
- L24 ANSWER 12 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Organopolysiloxanes

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L24 ANSWER 13 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN
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TI Isocyanatoaryl derivatives of phosphates and thiophosphates

L24 ANSWER 14 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Antibacterial compounds. I. Synthesis of several  $\mathbf{r}\mathbf{y}$  droxy chloro chalcones

L24 ANSWER 15 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Antispasmodics

L24 ANSWER 16 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI N-(2-Substituted-thiophenyl)-N-phenylalkylenediamines

L24 ANSWER 17 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Insecticides

L24 ANSWER 18 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Derivatives of 4-(diethylaminoethoxy)diphenylpropane

L24 ANSWER 19 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Quaternary ammonium salts of  $\delta\text{-hydrocarboxyphenyl-}\gamma\text{-hydroxy}$  amines

L24 ANSWER 20 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Derivatives of phenothiazine

#### => d 124 20 ti fbib abs

L24 ANSWER 20 OF 31 CAPLUS COPYRIGHT 2004 ACS on STN

TI Derivatives of phenothiazine

AN 1956:36301 CAPLUS

DN 50:36301

OREF 50:7153h-i,7154a

TI Derivatives of phenothiazine

PA Societe des usines chimiques de Rhone-Poulenc

DT Patent

LA Unavailable

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI GB 724217

19550216

GB

AB x, 10-RY(CH2)3R' (I) are prepared where x = 2 or 4, R = alkyl or alkoxy, Y = the phenothiazine nucleus (C.A. numbering), and R' = an amino group. To 10.6 g. 2-MeY, m. 187-8° (from S and 3-MeC6H4CH2Ph), 40 g. xylene, and 2.53 g. 90% NaNH2 is added 8 g. Me2N(CH2)3Cl in 8 g. xylene during 1 hr. at reflux, refluxing continued 1 hr., the mixture cooled, taken up in 150 cc. water, and slightly acidified with HCl, the organic layer decanted, the aqueous layer made strongly alkaline with caustic soda, and the base

extracted with ether and **distilled** to yield 11.6 g. 2,10-MeY(CH2)3NMe2, b0.1  $180-8^\circ$ ; HCl salt, m.  $194^\circ$ . I(x = 2, R = Me, R' = pyrrolidino) b0.15  $207-12^\circ$ ; oxalate, m.  $175^\circ$ . The following I are also reported (x = 2 or 4, but not further specified; R and R' given): MeO, pyrrolidino, b0.1  $205-7^\circ$  (HCl salt, m.  $143-5^\circ$ ); MeO, Me2N, b0.1  $198-202^\circ$  (HCl salt, m.  $170-95^\circ$ ); EtO, Me2N, b0.15  $212-16^\circ$  (oxalate, m.  $157-8^\circ$ ); BuO, Me2N, b0.15  $202-5^\circ$  (oxalate, m.  $154^\circ$ ); PrO, Me2N, b0.15  $211^\circ$  (oxalate, m.  $178-9^\circ$ ; MeI salt, m.  $70^\circ$ ). These compds. possess specific advantage as potentiators of general and local anesthetics and of analgesics. Cf. following abstract

=> d 126 1-7 ti

- L26 ANSWER 1 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Dynamics within a yard-waste compost biofilter during a ammonia biotransformation
- L26 ANSWER 2 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI The involvement of smooth muscle P2X receptors in the prolonged vasorelaxation response to purine nucleotides in the rat mesenteric arterial bed
- L26 ANSWER 3 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Stabilization of purified trimethoxysilane by storing in metal container
- L26 ANSWER 4 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Procedure for the preparation of proton-free deuterated chloroform
- L26 ANSWER 5 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Minimum reflux ratio and possibility of energy saving on a plate-to-plate heat-integrated **distillation** column
- L26 ANSWER 6 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Pulses of electrohydrodynamic sonoluminescence accompanying a high-voltage electric discharge in water
- L26 ANSWER 7 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Batch fractionation. Calculation of theoretical plates required for separation of two normal liquids

# => d 126 5 ti fbib abs

- L26 ANSWER 5 OF 7 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Minimum reflux ratio and possibility of energy saving on a plate-to-plate heat-integrated **distillation** column
- AN 1988:77845 CAPLUS
- DN 108:77845
- TI Minimum reflux ratio and possibility of energy saving on a plate-to-plate heat-integrated **distillation** column
- AU Nakaiwa, Masaru; Owa, Masaru; Akiya, Takaji; Kawasaki, Shigetake; Sato, Masahito; Lueprasitsakul, Veerapot; Takamatsu, Takeichiro
- CS Natl. Chem. Lab. Ind., Tsukuba, 305, Japan
- SO Sekiyu Gakkaishi (1988), 31(1), 81-6 CODEN: SKGSAE; ISSN: 0582-4664
- DT Journal
- LA Japanese
- AB A plate-to-plate heat-integrated **distillation** column, with a compressor and a throttling valve between the rectifying section and the stripping section and heat exchange between these sections, is considered. The rectifying section is operated at a higher temperature than the stripping section by compressing the vapor from the stripping section. The total vapor flow and liquid flow rates decrease gradually with heat exchange, toward the top in the rectifying section and toward the bottom in the stripping section. The external reflux ratio is less than the min. reflux ratio of a conventional column. The min. reflux was determined theor. The amount of energy required for the separation was estimated by simulating the

model

system. The states of min. reflux were determined in 3 conditions and the possible amount of energy reduction was obtained by calculating the loads on the

reboiler and the condenser from the external reflux ratio and on the compressor from the reflux ratio (RMAX) at the bottom stage of the rectifying section and operating pressures of the two sections. The min. value of the compressor load can be obtained from the min. condition of RMAX.

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FULL ESTIMATED COST	190.70	190.91
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CA SUBSCRIBER PRICE	-22.18	-22.18

SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 09:34:59 ON 26 FEB 2004